

Volume 40 Issue 2 December 2009 Pages 75-83 International Scientific Journal published monthly by the World Academy of Materials and Manufacturing Engineering

# Fabrication and characterization of TiAI/Ti<sub>3</sub>AI-based intermetallic composites (IMCs) reinforced with ceramic particles

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Received 25.09.2009; published in revised form 01.12.2009

#### ABSTRACT

**Purpose:** The purpose of the paper is to fabricate and characterise TiAl/Ti<sub>3</sub>Al-based intermetallic composites (IMCs) reinforced with ceramic particles.

**Design/methodology/approach:** Composites were formulated by blending commercially available powders of either TiAl or Ti<sub>3</sub>Al (technical grade with traces of Al and Ti) with ceramic powders ( $B_4C$ , TiC or TiB<sub>2</sub>) in appropriate amounts to create titanium aluminide-based matrices with 10, 20, 30, 40 and 50 vol. % of B<sub>4</sub>C, TiC or TiB<sub>2</sub> discontinuous reinforcement. The powder blends were thoroughly mixed and subsequently cold compacted.

**Findings:** Qualitative metallographic analysis of the as-densified microstructures confirmed that during densification of the composite matrix both TiAl and Ti<sub>3</sub>Al single phase titanium aluminide powders were transformed into various intermetallic phases (TiAl, Ti<sub>3</sub>Al and TiAl<sub>3</sub>). Regarding the room temperature tensile properties, excellent tensile strength, tensile yield strength and Young's modulus were measured in all fully dense composite samples, irrespective of their phase composition and volume fraction of reinforcement.

**Research limitations/implications:** Generally, the improvement of tensile strength, tensile yield strength and Young's modulus was found to be correlated with increase in the amount of ceramic reinforcement in the matrix. However, quite the opposite behaviour was found regarding elongation, where the introduction of ceramic particles into the intermetallic matrix led in all specimens to a significant reduction of elasticity.

**Originality/value:** In all systems and compositions, fully dense composite samples (with a retained porosity less than 1 vol. %) were successfully obtained, revealing the significant industrial potential of this fabrication method.

#### Keywords: Composites; TiAl

#### Reference to this paper should be given in the following way:

V. Kevorkijan, S.D. Škapin, Fabrication and characterization of TiAl/Ti<sub>3</sub>Al-based intermetallic composites (IMCs) reinforced with ceramic particles, Archives of Materials Science and Engineering 40/2 (2009) 75-83.

MATERIALS

### **1. Introduction**

TiAl/Ti<sub>3</sub>Al-based alloys have several advantages over conventional titanium alloys, such as higher elastic modulus,

lower density, better mechanical properties at elevated temperatures, and higher oxidation resistance due to the formation of a surface-passivated alumina layer [1,2]. TiAl/Ti<sub>3</sub>Al intermetallic-matrix composites (IMCs) reinforced with ceramic particles generally possess even higher specific strength and

specific stiffness, increased creep strength, improved toughness, and high-temperature strength retention [3, 4].

However, bringing these attractive intermetallic composite matrices into commercial use largely depends upon availability of practical and competitive enough processing routes. Due to difficulties in their production by conventional methods and high cost of powder processing, the elemental powder metallurgy (EPM) route has been gaining more and more attention because near-net shape titanium aluminide alloy products can be fabricated by the consolidation and forming of blended Ti and Al elemental powders, followed by a subsequent pressureless reactive synthesis and sintering process [5-11]. However, due to large difference between the partial diffusion coefficients of Ti and Al and the immobility of Ti atoms towards Ti/Al solid state reaction, the synthesis of TiAl/Ti<sub>3</sub>Al alloys via reactive sintering follows a mechanism in which Al atoms move into the Ti lattice, thus leading to the formation of Kirkendall diffusion pores [10, 12, 13]. Although hot isostatic pressing (HIP) and other pressure assisted methods have been reported to be effective in eliminating the porosity of reactively sintered TiAl/Ti<sub>3</sub>Al alloys [14-19], high cost and low production efficiency make it unsuitable for commercial use.

However, in the present study the assumption was made that the appearance of Kirkendall diffusion pores during pressureless sintering may be significantly reduced by replacing Ti and Al elemental powders with TiAl/Ti<sub>3</sub>Al, previously synthesized via EPM. To confirm this hypothesis, the experimental route consisted of pressureless reactive synthesis and then a pressureless sintering process performed in two separate stages in which in the first step the EMP technique was applied to obtain TiAl/Ti<sub>3</sub>Al powders from which, in the second step, fully dense IMCs reinforced with ceramic particles were fabricated.

Thus, the aim of this study (the first part of a research project) was to investigate the potential of the pressureless sintering method in fabrication of fully dense, high quality  $TiAl/Ti_3Al$  - based IMCs by applying reaction mixtures consisting of commercial titanium aluminide powders doped with various amounts (from 10 to 50 vol. %) of different ceramic particles (TiB<sub>2</sub>, TiC, B<sub>4</sub>C).

### 2. Phase diagrams

Examination of the Ti-Al phase diagram, Fig. 1 [20], shows presence of the following compounds in this system; Ti<sub>3</sub>Al (super alpha,  $\alpha_2$ ), TiAl (gamma,  $\gamma$ ), TiAl<sub>2</sub> (delta,  $\delta$ ) and TiAl<sub>3</sub>, and  $\alpha$ -Ti,  $\beta$ -Ti and  $\alpha$ -Al terminal solid solutions.

 $Ti_3Al$  has the tetragonal  $DO_{19}$  structure and TiAl has the hexagonal  $LI_0$  structure (which is basically an fcc lattice with atomic ordering and tetragonal distortion). The trialuminide  $TiAl_3$  crystallizes with the tetragonal  $DO_{22}$  structure.

For structural applications the near gamma  $(\alpha_2+\gamma)$  phase is more reliable. Near gamma titanium aluminides are classified into single and two phase Ti-(46-52) at% Al alloys.

The single phase near gamma alloys exhibit a lamellar microstructure. On the other hand, the significant two phases near gamma alloy appears in two different structures depending on the aluminium content. In the case of Ti-(46-48) at% Al the structure is a nearly lamellar and a duplex for Ti-(49-52) at% Al. Nearly

lamellar structures are defined as coarse lamellae grains with minor amounts of fine  $\gamma$  grains or gamma grains in the lamellar grain boundaries. The lamellar structure consists of alternating plates of the  $\gamma$ -TiAl and  $\alpha_2$ -Ti<sub>3</sub>Al phases. Such a lamellar structure results from the solid state phase transformation of the primary disordered  $\alpha$  dendrites. The  $\gamma$  regions surrounding the lamellar grains result from the transformation of the aluminium-rich interdendritic melt.



Fig. 1 .Ti-Al phase diagram [20]

#### **3. Experimental**

Composites were formulated by blending commercially available powders of either TiAl or Ti<sub>3</sub>Al (technical grade with traces of Al and Ti) with ceramic powders ( $B_4C$ , TiC or TiB<sub>2</sub>) in appropriate amounts to create titanium aluminide-based matrices with 10, 20, 30, 40 and 50 vol. % of  $B_4C$ , TiC or TiB<sub>2</sub> discontinuous reinforcement.

The powder blends were thoroughly mixed and subsequently cold compacted. In all cases, the reaction synthesis was conducted at  $1300^{\circ}$ C for 2 h in an Ar+4 vol. % H<sub>2</sub>-rich environment using a vacuum furnace.

The as-synthesized composite samples were cut, machined and polished in accordance with standard procedures.

Microstructural characterisation was performed by optical and scanning electron microscopy (OM and SEM), whereas X-ray diffraction (XRD) measurements were applied to the samples to identify the phases and their crystal structure.

The specimens for OM observation were subjected to electrolytic polishing in a solution of 95% CH<sub>3</sub>COOH and 5% HClO<sub>4</sub>, and then etched in a solution of 5% HNO<sub>3</sub>, 15% HF, and 80% H<sub>2</sub>O. The main grain sizes were measured by the linear intercept method.

The specimens for XRD were abraded with SiC paper and were then subjected to diffraction using  $CuK_{\alpha}$  radiation.

Quantitative determination of the volume percentage of ceramic particles in polished composite bars and the retained porosity was performed by analysing optical and scanning electron micrographs of infiltrated composites using the point counting method and image analysis and processing software. Composite density measurements were carried out in accordance with Archimedes' principle, applying distilled water as the immersion fluid.

The initial density of the green compacts was calculated from the mass and geometry of the samples.

The tensile properties (tensile strength, 0.2% tensile yield strength and elongation) of the composite specimens were determined in accordance with the ASTM test method, E8M-96. The tensile tests were conducted on round tension-test specimens 3.5 mm in diameter and 16 mm gauge length using an automated servo-hydraulic tensile testing machine with a crosshead speed of 0.254mm/60 s.

# 4. Results and discussion

# 4.1. Morphology of titanium aluminide powders applied

Morphology of titanium aluminide powders applied is shown in Figs. 2 and 3.





Fig. 2. SEM micrograph of as-received commercial TiAl powder





Fig. 3. SEM micrograph of as-received commercial Ti<sub>3</sub>Al powder

#### 4.2. Microstructure development in IMCs

Generally, the microstructure of IMCs consists of an intermetallic matrix (based on an ordered intermetallic compound or a multiphase combination of intermetallic compounds), the ceramic particulate reinforcement and the interfacial region with secondary phases formed during reactive sintering.

Regarding the matrix itself, during the last three decades a large variety of titanium aluminide microstructures was developed. The most promising microstructures are generally referred to as fully-lamellar, nearly-lamellar and duplex. The fully-lamellar microstructure consists of colonies (or grains) containing aligned platelets (lamellae) of  $\gamma$ -TiAl and  $\alpha_2$ -Ti<sub>3</sub>Al intermetallic phases [21]. In the case of the nearly-lamellar microstructure, the boundaries of  $\gamma$ -TiAl +  $\alpha_2$ -Ti<sub>3</sub>Al colonies, and, in particular, three-colony junctions are decorated with fine  $\gamma$ -TiAl grains [21].

The duplex microstructure consists of a mixture of equiaxial  $\gamma$ -TiAl grains and  $\gamma$ -TiAl +  $\alpha_2$ -Ti<sub>3</sub>Al colonies [21].

The microstructure of individual  $\gamma$ -TiAl+  $\alpha_2$ -Ti<sub>3</sub>Al colonies consists of parallel  $\gamma$ -TiAl and  $\alpha_2$ -Ti<sub>3</sub>Al lamellae with the lamellar

orientation varying from colony to colony. The  $\alpha_2$ -Ti<sub>3</sub>Al phase is generally the minor phase whose volume fraction is typically less than 20 vol%. The lamellae of the two phases generally do not alternate. Rather, several lamellae of the crystallographicalyrelated variants of the  $\gamma$ -TiAl phase separate the adjacent  $\alpha_2$ -Ti<sub>3</sub>Al lamellae. The  $\beta$ -phase particles are, in general, preferentially located at three-colony junctions [22].

SEM micrographs of various reactive sintered composite samples are shown in Figs. 4-15. A near-uniform distribution of the ceramic particulate reinforcements ( $B_4C$ ,  $TiB_2$  and TiC) through the both TiAl and  $Ti_3Al$  intermetallic matrices with no distinct evidence of clustering, or agglomeration is observed. This proves the effectiveness of the reactive sintering for producing composite microstructures.

In both TiAl and Ti<sub>3</sub>Al intermetallic matrices, inclusions of Ti particles and solidified aluminium, were occasionally found.

As a rule, in samples based on TiAl matrix, the traces of  $Ti_2Al_5$ ,  $TiAl_2$ ,  $Ti_5Al_{11}$  and  $Ti_9Al_{23}$  phases was observed. The suggested mechanism of the above phases formation is through a series of solid-liquid and/or solid state reactions necessarily involving TiAl as one of the starting phases [23].

An additional phase, frequently detected in both TiAl and  $Ti_3Al$ -based composites, is  $TiAl_3$ . This phase is formed during reactive sintering as the product of the reaction between solid Ti and liquid aluminium [23]:

$$\mathrm{Ti}_{(\mathrm{s})} + \mathrm{Al}_{(\mathrm{l})} = \mathrm{Ti}\mathrm{Al}_{3(\mathrm{s})} \tag{1}$$

As reported by Sujata et al. [23],  $TiAl_3$  is the only product of Reaction 1, because other possible products (TiAl and  $Ti_3Al$ ) are with significantly higher free energies of formation.

# **4.3. B**<sub>4</sub>**C**-titanium aluminide (TiAl and Ti<sub>3</sub>Al) system

Reactive sintering of  $B_4C$ -TiAl and  $B_4C$ -Ti<sub>3</sub>Al composite samples resulted in specimens with densities higher than 95% of T.D. In common with densification, boron carbide particulate reinforcement reacted with intermetallic matrix forming secondary ceramic reinforcing phases (TiB<sub>2</sub>, TiC, Al<sub>2</sub>Ti<sub>4</sub>C<sub>2</sub>, Al<sub>3</sub>BC), usually detected in sintered samples at the B<sub>4</sub>C-matrix interface.

In the systems  $B_4C$ -Ti<sub>3</sub>Al and  $B_4C$ -TiAl intensive chemical reactivity between the composite constituents was observed during high temperature densification due to the fact that boron carbide reacts with both molten aluminium and solid titanium [24, 25].

Boron carbide reacts at high temperature  $(1300^{\circ}C)$  with aluminium, forming mainly  $AlB_{24}C_4/Al_4C_3$  and some  $AlB_{12}$  (Reaction 2):

$$21B_4C_{(s)} + 16Al_{(l)} = 3AlB_{24}C_{4(s)} + 3Al_4C_{3(s)} + AlB_{12}$$
(2)

However, boron carbide also reacts with solid titanium particles producing secondary TiC and TiB<sub>2</sub>:

 $B_4C_{(s)} + 3Ti_{(s)} = 2TiB_{2(s)} + TiC_{(s)}$ (3)

The possible overall stoichiometric reactions between boron carbide and particular titanium aluminides used in experimental work are the following:

 $\begin{array}{l} 32B_4C_{(s)} + 16Ti_3Al = 32TiB_{2(s)} + 16TiC_{(s)} + 3AlB_{24}C_{4(s)} + 3Al_4C_{3(s)} \\ + AlB_{12(s)} \end{array} \tag{4}$ 

 $79B_4C_{(s)} + 48TiAl = 32TiB_{2(s)} + 16TiC_{(s)} + 9AlB_{24}C_{4(s)} + 9Al_4C_{3(s)} + 3AlB_{12(s)}$ (5)



Fig. 4. SEM micrograph of fully dense AlTi-10%  $B_4C$  composite samples with (1) an AlTi alloy matrix (fine and equiaxed  $\gamma$ -TiAl with traces of Ti<sub>2</sub>Al<sub>5</sub>, TiAl<sub>2</sub>, Ti<sub>5</sub>Al<sub>11</sub> and Ti<sub>9</sub>Al<sub>23</sub>), (2) TiB<sub>2</sub> bright grains, (3) TiC gray grains and (4) Al<sub>3</sub>Ti gray phase in the matrix



Fig. 5. SEM micrograph of a reactively sintered TiAl-10 vol. %  $B_4C$  composite showing different phases: (1)  $Al_3Ti$ , (2) TiB2 and (3)  $Al_2Ti_4C_2$ 

# 4.4. TiB<sub>2</sub>-titanium aluminide (TiAl and Ti<sub>3</sub>Al) system

 $TiB_2$  is stable at 1300°C in contact with a TiAl and  $Ti_3Al$  matrix. The main microstructural features that can be

varied in that case include the matrix grains morphology and distribution.



Fig. 6. Representative microstructures of a fully dense  $Ti_3Al-40$  vol. %  $B_4C$  composite prepared by reactive sintering of  $Ti_3Al + 40$  vol. %  $B_4C$ : (1) dark spots- $B_4C$ , (2) and (3) bright phases- $AlB_2$  and  $TiB_2$ , (4) and (5) grey phases-Al and probably also  $Al_4C_3$ . In  $Ti_3Al$  matrix, traces of  $Al_3BC$  are also identified



Fig. 7. Microstructure of TiAl-20 vol. %TiB<sub>2</sub> composite. The microstructural constituents are: (1) AlTi lamellae grey phase in the matrix with planar grain boundaries, (2) Ti<sub>3</sub>Al grains in and (3) TiB<sub>2</sub> grains

# 4.5. TiC-titanium aluminide (TiAl and Ti<sub>3</sub>Al) system

In this system, the best sintering results were experimentally obtained. Almost all the sintered samples, including these with a high amount (50 vol. %) of ceramic reinforcement, had a retained porosity below 5 vol. %, beside which numerous were fully dense.

Regarding high temperature  $(1300^{\circ}C)$  reactivity in TiAl-TiC and Ti<sub>3</sub>Al samples, Ti<sub>3</sub>AlC and Al<sub>3</sub>C<sub>3</sub> phases are confirmed at the matrix-reinforcement interface, according to the following chemical reactions [26]:

$$\begin{array}{l} 9\text{TiC}_{(s)} + 11\text{Al}_{(l)} = 2\text{Al}_4\text{C}_{3(s)} + 3\text{Ti}_3\text{AlC}_{(s)} & (6) \\ \text{or} & \\ 3\text{TiC}_{(s)} + 13\text{Al}_{(l)} = \text{Al}_4\text{C}_{3(s)} + 3\text{Ti}\text{Al}_{3(s)} & (7) \end{array}$$

Thereby, as in the case of the other reactive systems (TiAl-

 $B_4C$  and  $Ti_3Al-B_4C$ ) considered in this work, in TiAl-TiC and  $Ti_3Al$ -TiC, a variety of secondary microstructural features was also found.



Fig. 8. Scanning electron micrographs of  $Ti_3Al-20$  vol. %  $TiB_2$  composite showing: (1)  $Al_3Ti$  in sintered  $Ti_3Al$  matrix reinforced with (2)  $TiB_2$  grains



Fig. 9. Microstructure of a TiAl-30 vol. % TiB<sub>2</sub> composite showing different morphologies of the phases observed in the composite: (1) Al<sub>3</sub>Ti grains in TiAl matrix, (2) TiB<sub>2</sub> grains, (3) Al<sub>2</sub>O<sub>3</sub> dark inclusions and (4) traces of Ti<sub>2</sub>Al<sub>5</sub>, TiAl<sub>2</sub>, Ti<sub>5</sub>Al<sub>11</sub> and Ti<sub>9</sub>Al<sub>23</sub>



Fig. 10. SEM micrograph of the  $Ti_3Al$ -10 vol. % TiC composite with: (1)  $Ti_3Al$  grains in matrix, (2) a needle-like  $Ti_3AlC$  and (3) an Al-based phase



Fig. 13. SEM micrograph of the AlTi-30 vol. %TiC composite. The individual phases are: (1)  $\gamma$ -TiAl and  $\alpha_2$ -Al<sub>3</sub>Ti colonies, (2) Ti-Al-C based grains, (3) Ti<sub>3</sub>AlC and (4) Al-based phase



Fig. 11. Microstructure developed in Ti3Al-20 vol. %TiC



Fig. 12. Individual phases in the microstructure of a  $Ti_3Al-20$  vol.% TiC composite: (1) Ti-Al phase (2)  $Al_3Ti$ , (3) Al-based phase, (4) TiC and (5)  $Ti_2AlC$ 



Fig. 14. SEM micrographs of a fully dense  $\rm Ti_3Al\mathchar`alpha\mbox{-}40$  vol. % TiC composite



Fig. 15. Individual phases in a fully dense  $Ti_3Al-40$  vol. % TiC composite: (1)  $Ti_3Al$ , (2) Ti-Al-C based grains, (3)  $Ti_2AlC$  and (4) Al-based phase

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Average room temperature tensile properties of various laboratory prepared composite samples

No.	Initial chemical composition (vol. %)	Retained porosity (%)	E (GPa)	Tensile strength (MPa)	0.2 % tensile yield strength (MPa)	Elongation in 50 mm (%)
	90% Ti <sub>3</sub> Al + 10% TiB <sub>2</sub>	1.4±0.1	268±25	721±70	442±40	0.8±0.08
	90% Ti <sub>3</sub> Al + 10% B <sub>4</sub> C	1.2±0.1	257±25	698±70	397±40	0.7 ±0.07
	90% Ti <sub>3</sub> Al + 10% TiC	1.1±0.1	298±30	774±80	483±50	0.8±0.08
	90% TiAl + 10% TiB <sub>2</sub>	1.5±0.2	196±20	726±70	541±50	0.8±0.08
	90% TiAl + 10% B <sub>4</sub> C	1.4±0.1	192±20	698±70	518±50	0.7±0.07
	90% TiAl + 10% TiC	1.4±0.1	208±20	739±70	573±60	0.8±0.08
	$80\% Ti_3Al + 20\% TiB_2$	2.1±0.2	298±30	812±80	488±50	0.6±0.06
	80% Ti <sub>3</sub> Al + 20% B <sub>4</sub> C	1.8±0.2	287±20	768±80	445 <u>±</u> 40	0.7±0.07
	80% Ti <sub>3</sub> Al + 20% TiC	1.9±0.2	335±20	868±90	549±50	0.5±0.05
	80% TiAl + 20% TiB <sub>2</sub>	2.3±0.2	238±25	798±80	593±60	0.6±0.06
	80% TiAl + 20% B <sub>4</sub> C	1.9±0.2	218±20	771±80	567±60	0.7±0.07
	80% Ti <sub>3</sub> Al + 20% TiC	1.9±0.2	236±25	819±80	622±60	0.7±0.07
	70% $Ti_3Al + 30\% TiB_2$	2.5±0.3	340±30	893±90	539±50	0.6±0.06
	$70\% Ti_3Al + 30\% B_4C$	1.9±0.2	332±30	864±90	503±50	0.6±0.06
	70% Ti <sub>3</sub> Al + 30% TiC	2.1±0.2	351±35	947±90	589±60	0.6±0.06
	70% TiAl + 30% TiB <sub>2</sub>	3.0±0.3	281±40	901±90	641±60	0.5±0.05
	70% TiAl + 30% B <sub>4</sub> C	1.9±0.2	254±45	865±90	611±60	0.5±0.05
	70% TiAl + 30% TiC	1.9±0.2	287 <u>+</u> 45	913±90	686±70	0.6±0.06
	$60\% Ti_3Al + 40\% TiB_2$	3.1±0.3	376±45	997±100	598±60	0.4±0.04
	$60\% Ti_3Al + 40\% B_4C$	1.8±0.2	368±45	994±100	564±60	0.5±0.05
	60% Ti <sub>3</sub> Al + 40% TiC	1.9±0.2	392 <u>+</u> 45	1022±100	648±60	0.3±0.03
	60% TiAl + 40% TiB <sub>2</sub>	3.6±0.4	318±45	988±100	699±70	0.4±0.04
	60% TiAl + 40% B <sub>4</sub> C	2.9±0.3	288±45	959±100	671±70	0.4±0.04
	60% TiAl + 40% TiC	1.9±0.2	322±45	973±100	720±70	0.4±0.04
	50% $Ti_3Al + 50\% TiB_2$	4.2±0.4	440 <u>+</u> 45	1108±110	677±70	0.3±0.03
	$50\% Ti_3Al + 50\% B_4C$	2.8±0.3	448 <u>+</u> 45	1105±110	658±70	0.3±0.03
	50% Ti <sub>3</sub> Al + 50% TiC	2.9±0.3	453±45	1129±110	704±70	0.3±0.03
	50% TiAl + 50% TiB <sub>2</sub>	5.1±0.5	402 <u>+</u> 45	1103±110	768±80	0.1±0.01
	50% TiAl + 50% B <sub>4</sub> C	2.8±0.3	397±45	1087±110	759±80	0.2±0.02
	50% TiAl + 50% TiC	2.9±0.3	421±45	1116±110	799±80	0.1±0.01

#### 5. Mechanical properties

The results of the room temperature tensile tests on composite samples are listed in Table 1. As a result of matrix reinforcement, significant improvements in Young's modulus, tensile yield strength and ultimate tensile strength of the fabricated composites were observed, resulting in IMCs with excellent mechanical properties. These mechanical properties were found to be slightly higher in composites with a  $Ti_3Al$ -based matrix compared to TiAl-based matrix counterparts. At the same time, the highest improvement of tensile properties caused by reinforcement was observed in samples reinforced with TiC and TiB<sub>2</sub> and less in samples reinforced with B<sub>4</sub>C. Comparing the mechanical properties of composite samples with various volume fractions of ceramic particles in the matrix, it was found that Young's modulus, tensile yield strength and ultimate tensile strength increase while elongation decreases with an increasing fraction of ceramic reinforcement.

## **6.** Conclusions

A study of the fabrication of titanium aluminide-based intermetallic matrix composites (IMCs) discontinuously reinforced with various amounts (from 10 to 50 vol. %) of ceramic particles ( $B_4C$ , Ti $B_2$  and TiC) was conducted by applying conventional pressureless reactive sintering of single phase titanium aluminide powders (TiAl or Ti<sub>3</sub>Al) and ceramic reinforcement. In all systems and compositions, fully dense composite samples (with a retained porosity less than 1 vol. %) were successfully obtained, revealing the significant industrial potential of this fabrication method.

Qualitative metallographic analysis of the as-densified microstructures confirmed that during densification of the composite matrix both TiAl and Ti<sub>3</sub>Al single phase titanium aluminide powders were transformed into various intermetallic phases (TiAl, Ti<sub>3</sub>Al and TiAl<sub>3</sub>). In addition, numerous secondary phases formed *in situ* by chemical reactions between titanium aluminide phases and the ceramic reinforcements were also identified.

In the case of a titanium aluminide matrix reinforced with  $B_4C$ , secondary phases formed in situ were  $TiB_2$  and TiC, although traces of  $AlB_{24}C_4$ ,  $Al_4C_3$  and  $AlB_{12}$  were also confirmed. In composite titanium aluminides reinforced with  $TiB_2$  no appearance of secondary phases was found. However, intensive phase transformations occurred, particularly along the grain boundaries of the sintered titanium aluminide skeleton.

The highest chemical reactivity and also sinterability was observed in TiAl-TiC and Ti<sub>3</sub>Al-TiC samples. The dominant product was Ti<sub>3</sub>AlC with characteristic, needle-like morphology.

Regarding the room temperature tensile properties, excellent tensile strength, tensile yield strength and Young's modulus were measured in all fully dense composite samples, irrespective of their phase composition and volume fraction of reinforcement. The best values were obtained in TiAl-TiC and  $Ti_3Al$ -TiC samples with the highest amount (50 vol. %) of ceramic reinforcement. Generally, the improvement of tensile strength, tensile yield strength and Young's modulus was found to be correlated with the increase in the amount of ceramic reinforcement in the matrix.

However, quite the opposite behaviour was found regarding elongation, where the introduction of ceramic particles into the intermetallic matrix led in all specimens to a significant reduction of elasticity.

### Acknowledgements

This work was supported by funding from the Public Agency for Research and Development of the Republic of Slovenia, as well as the Impol Aluminium Company from Slovenska Bistrica, Slovenia, under contract No. 1000-07-219308.

### Additional information

The paper was published also in the Archives of Materials Science.

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